

Table1: Biochemical Reactions

Number	Reaction name	Reaction	Kinetic law	Flux (v)
1	Degradation of AHL by AiiA	$AiiA + AHL \rightarrow AiiA$	Michaelis - Menten	$v_1 = \frac{k_{1cat}[AiiA][AHL]}{K_{1M} + AHL}$
2	Complex formation and dissociation between AHL and LuxR	$AHL + LuxR \leftrightarrow AHL:LuxR$	Mass Action	$v_2 = k_2[AHL][LuxR] - k_{-2}[AHL:LuxR]$
2.1	Dimer formation and dissociation between AHL:LuxR complexes	$2 AHL:LuxR \leftrightarrow (AHL:LuxR):(AHL:LuxR)$	Mass Action	$v_{2.1} = k_{2.1}[AHL][LuxR]^2 - k_{-2.1}[(AHL:LuxR):(AHL:LuxR)]$
3.1	CI synthesis induced by AHL and LuxR complexes dimer	$\rho_{ci} + (AHL:LuxR):(AHL:LuxR) \rightarrow \rho_{ci} + (AHL:LuxR):(AHL:LuxR) + CI$	Mass Action	$v_{3.1} = k_{3on}[\rho_{ci}][(AHL:LuxR):(AHL:LuxR)]$
3.2	Constitutive CI synthesis	$\rho_{ci} \rightarrow \rho_{ci} + CI$	Mass Action	$v_{3.2} = k_{3off}[\rho_{ci}]$
4	Natural degradation of CI	$CI \rightarrow \emptyset$	Mass Action	$v_4 = k_4[CI]$
4.1	Dimer formation and dissociation between CI molecules	$2 CI \leftrightarrow CI:CI$	Mass Action	$v_{4.1} = k_{4.1}[CI]^2 - k_{-4.1}[CI:CI]$
6	RcnA production	$\rho \rightarrow \rho + RcnA$	Hill kinetics	$v_6 = \frac{k_6[\rho]}{1 + \frac{[CI:CI]}{k_{5.1}} + \frac{[CI:CI]}{k_{5.2}} + \frac{[CI:CI]^2}{k_5}}$
7	Nickel efflux by RcnA	$RcnA + Ni_{int} \rightarrow RcnA + Ni_{ext}$	Mass Action	$v_7 = k_7[RcnA][Ni_{int}]$
8	Natural degradation of RcnA	$RcnA \rightarrow \emptyset$	Mass Action	$v_8 = k_8[RcnA]$
9	Nickel import by unknown channel	$Unk + Ni_{ext} \rightarrow Unk + Ni_{int}$	Mass Action	$v_9 = k_9[Unk][Ni_{ext}]$